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Experiment 1: physical and chemical properties of

Objective: To study the chemical and physical properties of hydrocarbons. Background: In this experiment you will observe the solubility (a physical property) and the

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hydrocarbon, carbon: hydrocarbons any of a class of organic chemical compounds composed only of the elements carbon (C) and hydrogen (H). The carbon atoms join

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Laboratory Manual. Winter 2009. Table of Contents. Exp. # Title of Experiment. Page. Please refer to the instructions given in Lab 3 (Hydrocarbons) for the format.

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Appendix i - hazards

Organic compounds composed solely of carbon and hydrogen. Hydrocarbons may be classified into 3 broad categories: Open-chain aliphatic compounds

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Saturated versus unsaturated hydrocarbons c11-5-10

Given the following hydrocarbons, determine whether the compound listed is an alkane, alkene or alkyne and whether the hydrocarbon is saturated or unsaturated.

Hydrocarbon gas analysis - laboratory training

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Activity #2

The Hydrocarbons and Their Derivatives: Nomenclature, Physical/Chemical Properties, Activity #2
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